

5. Enzyme-Substrate Kinetics

(J.D. Murray, Volume 1, Chapter 6, Sections 6.1 - 6.3).

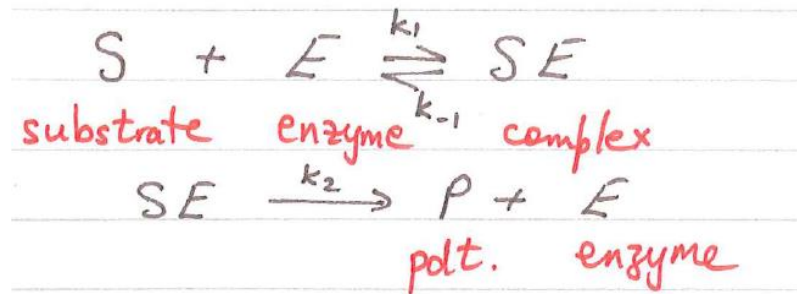
Summary of this presentation

- Derive the Michaelis-Menten model equations.

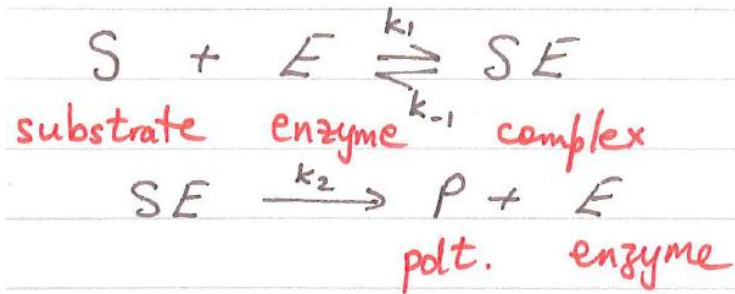
5.1 Michaelis-Menten (1913)

Consider the case of a substrate, S, combining with an enzyme, E, to produce a complex, SE, which then gives back the enzyme, together with the product, P.

Here, k_1, k_{-1}, k_2 are *rate constants*.



Law of mass action: The rate of a reaction is proportional to the product of the (active) concentrations of the reactants.



Let $s = [S]$, $e = [E]$, $c = [SE]$, $p = [P]$

$$\therefore \frac{ds}{dt} = -k_1 se + k_{-1} c$$

$$\frac{de}{dt} = -k_1 se + k_{-1} c + k_2 c$$

$$\frac{dc}{dt} = k_1 se - k_{-1} c - k_2 c$$

$$\frac{dp}{dt} = k_2 c \quad \leftarrow \text{uncoupled.}$$

N.B. $\frac{de}{dt} + \frac{dc}{dt} = 0 \Rightarrow e(t) + c(t) = e_0 \Rightarrow e(t) = e_0 - c(t)$

Conservation law – enzyme is conserved

I.c. $s(0) = s_0$, $e(0) = e_0$, $c(0) = 0$, $p(0) = 0$

$$\Rightarrow \frac{ds}{dt} = -k_1 e_0 s + (k_1 s + k_{-1}) c$$

$$\frac{dc}{dt} = k_1 e_0 s - (k_1 s + k_{-1} + k_2) c$$

Nondimensionalise: $\tau = k_1 e_0 t$, $u(\tau) = \frac{s(t)}{s_0}$, $v(\tau) = \frac{c(t)}{e_0}$, $\lambda = \frac{k_2}{k_1 s_0}$,

$$k = \frac{k_{-1} + k_2}{k_1 s_0}, \quad \varepsilon = \frac{e_0}{s_0}$$

$$\Rightarrow \quad u' = -u + (u + k - \lambda)v \quad u(0) = 1$$

$$\varepsilon v' = u - (u + k)v \quad v(0) = 0$$

Coupled system of two ODEs

- Do phase planes and linear stability analysis etc?
- Exercise

$$\lambda > 0, k > 0$$

$$\left[k - \lambda = \frac{k-1}{k_1 s_0} > 0 \right]$$

$$u' = -u + (u+k-\lambda)v \quad u(0) = 1$$

$$\varepsilon v' = u - (u+k)v \quad v(0) = 0$$

$$\therefore \text{Put } \varepsilon = 0 \quad \therefore v = \frac{u}{u+k}$$

$$\therefore u' = -u + \frac{(u+k-\lambda)u}{u+k} = -\frac{\lambda u}{u+k}$$

$$\Rightarrow u = u(\tau)$$

N.B. $v(0) = 0, u(0) = 1$

$$\therefore 0 = \frac{1}{1+k} \cdot \cancel{X} \cdot !!$$

This is a singular perturbation problem.

Summary

- Derived Michaelis-Menten Model.
- This is a singular perturbation problem.

End of Lecture 5-1

Summary of previous part

- Derived the Michaelis-Menten model equations
- Saw that it leads to a singular perturbation problem

Summary of this presentation

- Solve the singular perturbation problem
- Quasi-steady state assumption (Pseudo-steady state hypothesis)
- Matched asymptotic expansions

5.2 Singular Perturbation and the Quasi-Steady State Approximation (QSSA) or Pseudo-Steady State Hypothesis (PSSH)

$$\begin{aligned}u' &= -u + (u+k-\lambda)v & u(0) &= 1 \\ \varepsilon v' &= u - (u+k)v & v(0) &= 0\end{aligned}$$

Note, initially $u' \approx -1 \quad \therefore u$ decreasing
 $v' \approx \frac{1}{\varepsilon} \quad \therefore v$ increases rapidly

\therefore "expand" time near $t=0$ i.e. set $\sigma = \tau/\varepsilon$ Let $u(\tau, \varepsilon) = \hat{u}(\sigma, \varepsilon)$

$$\therefore \frac{d}{d\tau} = \frac{d}{d\sigma} \frac{1}{\varepsilon}$$

$$\frac{d\hat{u}}{d\sigma} = \varepsilon (-\hat{u} + (\hat{u}+k-\lambda)\hat{v})$$

$$\frac{d\hat{v}}{d\sigma} = \hat{u} - (\hat{u}+k)\hat{v}$$

Note that putting $\varepsilon = 0$ in the original equations turned the 2nd order system into a 1st order system. However, in this transformed system, we can put $\varepsilon = 0$ without changing the order of the system!

$$\begin{aligned}\hat{u} &= \hat{u}_0 + \varepsilon \hat{u}_1 + \dots & \hat{u}_i &= \hat{u}_i(\sigma) \\ \hat{v} &= \hat{v}_0 + \varepsilon \hat{v}_1 + \dots & \hat{v}_i &= \hat{v}_i(\sigma)\end{aligned}$$

$$\frac{d\hat{u}}{d\sigma} = \varepsilon (-\hat{u} + (\hat{u} + k - \lambda)\hat{v})$$

$$\frac{d\hat{v}}{d\sigma} = \hat{u} - (\hat{u} + k)\hat{v}$$

$$O(1): \frac{d\hat{u}_0}{d\sigma} = 0$$

$$\frac{d\hat{v}}{d\sigma} = \hat{u}_0 - (\hat{u}_0 + k)\hat{v}_0$$

$$\hat{u}(0) = 1, \hat{v}(0) = 0$$

$$\begin{aligned}\Rightarrow \hat{u}_0(\sigma) &= 1, \hat{v}_0(\sigma) = \frac{1}{1+k} [1 - e^{-(1+k)\sigma}] \\ &= \frac{1}{1+k} [1 - e^{-(1+k)\sigma/\varepsilon}]\end{aligned} \left. \vphantom{\begin{aligned}\Rightarrow \hat{u}_0(\sigma) &= 1, \hat{v}_0(\sigma) = \frac{1}{1+k} [1 - e^{-(1+k)\sigma}] \\ &= \frac{1}{1+k} [1 - e^{-(1+k)\sigma/\varepsilon}]\end{aligned}} \right\} \text{inner sol}^n.$$

Back to the original equations

- Let us assume that after the initial period of time in which v changes rapidly we can go back to the original system.

Outer solⁿ: $u(\tau) = u_0 + \varepsilon u_1 + \dots$
 $v(\tau) = v_0 + \varepsilon v_1 + \dots$

$$\frac{du}{d\tau} = -u + (u+k-\lambda)v$$

$$\varepsilon \frac{dv}{d\tau} = u - (u+k)v$$

$$O(1): \frac{du_0}{d\tau} = -u_0 + (u_0+k-\lambda)v_0$$

$$0 = u_0 - (u_0+k)v_0 \quad \text{QSSA or PSSH}$$

$$\Rightarrow v_0 = \frac{u_0}{u_0+k}$$

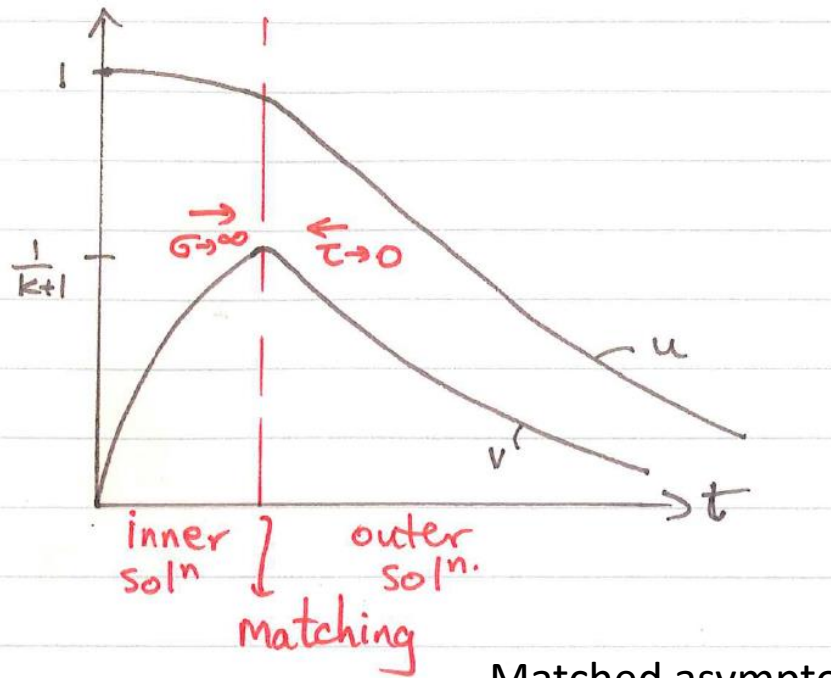
$$\Rightarrow u_0(\tau) + K \ln u_0(\tau) = A - \lambda \tau$$

$$u_0(0) = 1 \Rightarrow 1 = A$$

$$\tilde{v}_0(\sigma) = \frac{1}{1+k} [1 - e^{-(1+k)\sigma}]$$

$$= \frac{1}{1+k} [1 - e^{-(1+k)\tau/\varepsilon}]$$

$$\begin{aligned} \therefore u_0(\tau) + K \ln u_0(\tau) &= 1 - \lambda \tau \\ v_0(\tau) &= \frac{u_0(\tau)}{K + u_0(\tau)} \end{aligned} \quad \left. \vphantom{\begin{aligned} \therefore u_0(\tau) + K \ln u_0(\tau) &= 1 - \lambda \tau \\ v_0(\tau) &= \frac{u_0(\tau)}{K + u_0(\tau)} \end{aligned}} \right\} \text{outer sol}^n.$$



Matched asymptotic expansions

We have two timescales – a *fast* one and a *slow* one. v changes on the fast timescale (v is called the *fast variable*), u changes on the slow timescale (*slow variable*).

To investigate this problem analytically, we *separated* the timescales.

Usually the inner solⁿ is not seen \therefore we consider the outer solⁿ which, to 1st order, is given by

$$\frac{du}{dt} = -\frac{\lambda u}{u+K} \quad \text{Michaelis-Menten uptake.}$$

Defn. $r(u)$ is called the velocity of reaction if

$$\frac{du}{dt} = -r(u). \quad \frac{du}{d\tau} \text{ is called the } \underline{\text{uptake function.}}$$

- So, now we have solved (at least to lowest order) for the substrate and the complex.
- We can then integrate the equation for the product to find it.
- We can use the conservation law to find the enzyme concentration.

In general biochemical reactions occur due to several enzyme catalysts \therefore the rate eqns. have the

form:

$$\frac{du_1}{d\tau} = f(u_1, u_2, \dots, u_n)$$

$$\varepsilon_i \frac{du_i}{d\tau} = f_i(u_1, \dots, u_n), \quad i=2, 3, \dots, n \quad \varepsilon_i \ll 1 \quad \forall i$$

○ The PSSH assumes $f_i(u_1, \dots, u_n) = 0 \quad i=2, 3, \dots, n$

$$\Rightarrow u_i = u_i(u_1) \quad i=2, 3, \dots, n$$

$$\Rightarrow \frac{du_1}{d\tau} = f(u_1, h(u_1)) \quad \text{etc.}$$

Summary

- QSSA (PSSH)
- Asymptotics

End of Lecture 5-2