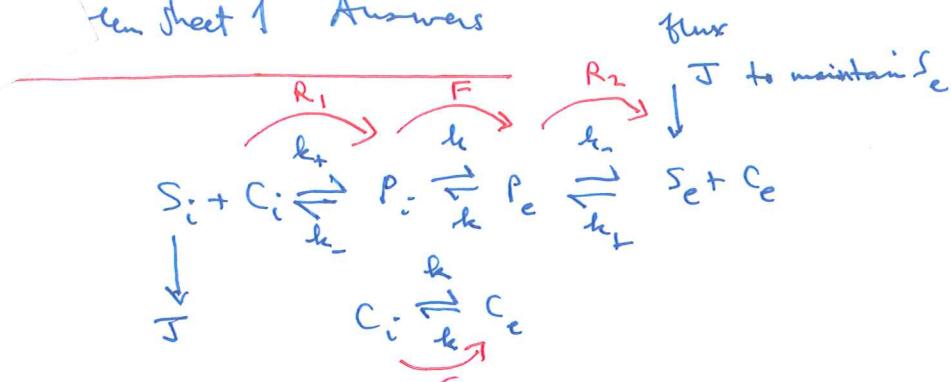


Sheet 1 Answers

1.



meaning, see notes. i, e internal, external binding sites

concentrations { S substrate binding to
C gate protein to form
P complex which switches conformationally
between i & e states.

The overall reaction rates are

$$R_1 = k_+ S_i C_i - k_- P_i$$

$$R_2 = k_- P_e - k_+ S_e C_e$$

$$F = k P_i - k P_e$$

$$G = k C_i - k C_e$$

then

$$\dot{S}_i = -R_1 - \star J$$

$$\dot{C}_i = -R_1 - G$$

$$\dot{P}_i = R_1 - F$$

$$\dot{P}_e = F - R_2$$

$$\dot{S}_e = R_2 + \star J$$

$$\dot{C}_e = R_2 + G$$

including
fluxes

~~Eqns. for S & C don't add up~~

In equilibrium $J = -R_1 = +G = -R_2 = -F$

& two conservation laws (substrate & gates)

$$P_i + P_e + C_i + C_e = C_0$$

$$P_i + P_e + S_i + S_e = S_0$$

(2)

$$S_0 \quad J = k_- p_i - k_+ s_i c_i \quad (1)$$

$$= k c_i - k c_e \quad (2)$$

$$= k_+ s_e c_e - k_- p_e \quad (3)$$

$$= k p_e - k p_i \quad (4)$$

$$S_0 \quad p_e = p_i + \frac{J}{k} \quad \text{for } \sim (4)$$

$$c_e = c_i - \frac{J}{k} \quad \sim (2)$$

$$\text{Substitute } 2(p_i + c_i) = C_0 \Rightarrow p_i = \frac{C_0}{2} - c_i$$

$$\& \quad k_- p_i - k_+ s_i c_i = J \quad \text{from (1)}$$

$$\Rightarrow k_- \left(\frac{C_0}{2} - c_i \right) - k_+ s_i c_i = J$$

$$\Rightarrow c_i = \frac{k_- \frac{C_0}{2} - J}{k_- + k_+ s_i} \quad \sim (1)$$

$$\text{use (3)} : \quad J = k_+ s_e \left[c_i - \frac{J}{k} \right] - k_- \left[\frac{C_0}{2} - c_i + \frac{J}{2k} \right]$$

$$\Rightarrow \cancel{k_+ s_e} J \left[1 + \frac{k_+ s_e}{4k} + \frac{k_-}{k} \right] = -\frac{k_- C_0}{2} + (k_- + k_+ s_e) \left[\frac{k_- \frac{C_0}{2} - J}{k_- + k_+ s_i} \right]$$

$$\text{Define } K_m = \frac{k_- + k_+}{k_+} \quad K_d = \frac{k_-}{k_+}$$

$$\Rightarrow J \left[\frac{k_+ (K_m + s_e)}{k} \right] = -\frac{k_- C_0}{2} + \left(\frac{k_- + k_+ s_e}{k_- + k_+ s_i} \right) \left[\frac{k_- \frac{C_0}{2} - J}{k_- + k_+ s_i} \right]$$

$$\text{note } \frac{k_-}{k_+} = K_m - K_d, \text{ so } = -\frac{k_- C_0}{2} + \left[\frac{K_m - K_d + s_e}{K_m - K_d + s_i} \right] \left(\frac{k_- \frac{C_0}{2} - J}{K_m - K_d + s_i} \right)$$

3

So

$$J \left[\frac{\frac{K_m + S_e}{K_d}}{\frac{K_m - K_d + S_i}{K_m - K_d + S_i}} \right] = \frac{k_c - C_0}{2} \left[-1 + \frac{K_m - K_d + S_e}{K_m - K_d + S_i} \right] + \frac{k_c C_0}{2}$$

$$- \left[\frac{K_m - K_d + S_e}{K_m - K_d + S_i} \right] J$$

$$J \left[\frac{\frac{K_m + S_e}{K_d}}{\frac{K_m - K_d + S_e}{K_m - K_d + S_i}} + \frac{\frac{K_m - K_d + S_e}{K_m - K_d + S_i}}{\frac{K_m - K_d + S_i}{K_m - K_d + S_i}} \right] = \frac{k_c - C_0}{2} \frac{(S_e - S_i)}{K_m - K_d + S_i}$$

$$J \left[\frac{K_m - K_d}{K_m} K_m + K_m S_i + (K_m - K_d) S_e + S_i S_e + K_d S_e - K_d^2 + K_d K_m \right] = \frac{k_c - C_0}{2} k_d (S_e - S_i)$$

$$\downarrow$$

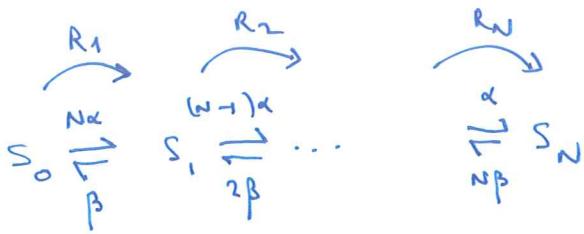
$$= J \left[K_m + K_m S_i + K_m S_e + S_i S_e - K_d^2 \right] = \frac{k_c - C_0}{2 k_c} (S_e - S_i)$$

so $J = \frac{\frac{k_c - C_0}{2 k_c} (S_e - S_i)}{\frac{(K_m + S_i)(K_m + S_e) - K_d^2}{2 k_c}}$

ungleichbar...

2.

(4)



Define R_i as shown, thus

$$R_1 = N\alpha S_0 - \beta S_1$$

$$R_2 = (N-1)\alpha S_1 - 2\beta S_2$$

$$R_i = (N-i+1)\alpha S_{i-1} - i\beta S_i$$

$$R_N = \alpha S_{N-1} - N\beta S_N$$

$$\text{and } S_0 = -R_1$$

$$S_1 = R_1 - R_2$$

$$S_i = R_i - R_{i-1}$$

$$S_{N-1} = R_{N-1} - R_N$$

$$S_N = R_N$$

$$\text{Conservation } S_0 + S_1 + \dots + S_N = 1$$

~~$$S_0 = \frac{N\alpha S_0}{\beta}$$~~

$$\text{Ansatz } S_j = \frac{N!}{(N-j)!j!} n^j (1-n)^{N-j}$$

$$k_j = (N-j+1) \times \frac{N!}{(N-j+1)!(j-1)!} n^{j-1}(1-n)^{N-j+1} - \frac{j\beta N!}{(N-j)!j!} n^j (1-n)^{N-j}$$

$$= \frac{N! \times n^{j-1}(1-n)^{N-j+1}}{(N-j)!(j-1)!} - \frac{\beta N!}{(N-j)!(j-1)!} n^j (1-n)^{N-j}$$

$$= \frac{N! n^{j-1}(1-n)^{N-j}}{(N-j)!(j-1)!} [\alpha(1-n) - \beta n]$$

(5)

$$\begin{aligned}
 \text{Note } \dot{s}_j &= \frac{n!}{(n-j)! \cdot j!} \left[j n^{j-1} (1-n)^{n-j} - (n-j) n^j (1-n)^{n-j-1} \right] : \\
 &= \frac{n!}{(n-j)! (j-1)!} n^{j-1} (1-n)^{n-j} : \\
 &\quad - \frac{n!}{(n-j-1)! \cdot j!} n^j (1-n)^{n-j-1} :
 \end{aligned}$$

$$\text{recall } R_j = \frac{n!}{(n-j)! (j-1)!} n^{j-1} (1-n)^{n-j} [\alpha(1-n) - \beta n]$$

$$\begin{aligned}
 \text{thus } \dot{s}_j &= \frac{R_j n}{[\alpha(1-n) - \beta n]} - \frac{R_{j+1} n}{[\alpha(1-n) - \beta n]} \\
 &= (R_j - R_{j+1}) \frac{n}{[\alpha(1-n) - \beta n]} \\
 &= R_j - R_{j+1}
 \end{aligned}$$

$$\therefore \dot{s}_j = \frac{n!}{(n-j)! \cdot j!} n^j (1-n)^{n-j}$$

$$\text{then } \underline{\underline{n}} = \alpha(1-n) - \beta n$$

(6)

For $N = 2$

$$\begin{aligned}\dot{s}_0 &= -R_1 & R_1 &= 2s_0 - \beta s_1 \\ \dot{s}_2 &= R_2 & R_2 &= \alpha s_1 - 2\beta s_2 \\ && \text{And } s_0 + s_1 + s_2 &= 1\end{aligned}$$

$$\begin{aligned}s_0 \quad \dot{s}_0 &= -2\alpha s_0 + \beta [1 - s_0 - s_2] \\ \dot{s}_2 &= \alpha [1 - s_0 - s_2] - 2\beta s_2\end{aligned}$$

$$\text{or} \quad \begin{aligned}\dot{s}_0 &= \beta - (2\alpha + \beta)s_0 - \beta s_2 \\ \dot{s}_2 &= \alpha - \alpha s_0 - (\alpha + 2\beta)s_2\end{aligned}$$

We already know $s_0 = \frac{(1-n)^2}{\alpha}$, $s_2 = \frac{n^2}{(\alpha+2\beta)}$ is a solution with $n = \alpha(1-n) - \beta n$

$$\text{Thus linearising} \quad s_0 = (1-n)^2 + y_0 \quad s_2 = \frac{n^2}{(\alpha+2\beta)} + y_2$$

$$\Rightarrow \begin{aligned}\dot{y}_0 &= - (2\alpha + \beta)y_0 - \beta y_2 \\ \dot{y}_2 &= -\alpha y_0 - (\alpha + 2\beta)y_2\end{aligned}$$

$$\begin{pmatrix} \dot{y}_0 \\ \dot{y}_2 \end{pmatrix} = - \underbrace{\begin{pmatrix} 2\alpha + \beta & \beta \\ \alpha & \alpha + 2\beta \end{pmatrix}}_{M}$$

$$\det(-M) = \det M = 2(\alpha + \beta)^2 > 0$$

$$\text{tr}(-M) = -3(\alpha + \beta) < 0$$

$$\text{Solutions } \alpha e^{\lambda t} \quad \lambda^2 + 3(\alpha + \beta) + 2(\alpha + \beta)^2 = 0$$

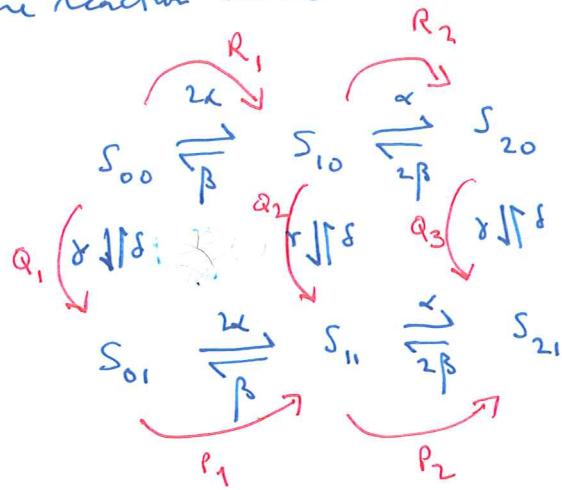
$$\lambda = -2(\alpha + \beta), -(\alpha + \beta)$$

\Rightarrow Stable.

3. [optimal depending on time]

(7)

The reaction scheme is



Reaction rates are $R_1 = 2\alpha S_{00} - \beta S_{10}$

$$R_2 = \alpha S_{10} - 2\beta S_{20}$$

$$Q_1 = \gamma S_{00} - \delta S_{01}$$

$$Q_2 = \gamma S_{10} - \delta S_{11}$$

$$Q_3 = \gamma S_{20} - \delta S_{21}$$

$$P_1 = 2\alpha S_{01} - \beta S_{11}$$

$$P_2 = \alpha S_{11} - 2\beta S_{21}$$

If then

$$\dot{S}_{00} = -R_1 - Q_1$$

$$\dot{S}_{10} = R_1 - R_2 - Q_2$$

$$\dot{S}_{20} = R_2 - Q_3$$

$$\dot{S}_{01} = Q_1 - P_1$$

$$\dot{S}_{11} = P_1 - P_2 + Q_2$$

$$\dot{S}_{21} = P_2 + Q_3$$

$$\sum S_{ij} = 1$$

Suppose $S_{21} = m^2 h$
 $S_{11} = 2m(1-m)h$
 $S_{20} = m^2(1-m)$

$$\left. \begin{array}{l} P_2 = 2am(1-m)h - \frac{2\beta m^2 h}{2\beta m^2 h} \\ Q_3 = \gamma m^2(1-m) - \delta m^2 h \end{array} \right\}$$

$$Q_3 = \gamma m^2(1-m) - \delta m^2 h$$

(8)

Density of open channels as gates are independent & an open channel needs $\frac{m}{M}$ open $\frac{h}{M}$ open & probability of M open is m , of M open is h

so prob of open channel is $m^2 h$

[A fair sample - we identify probability with a realisation]

$$\text{Put } S_{00} = (1-m)^2(1-h)$$

$$S_{10} = 2m(1-m)(1-h)$$

$$S_{20} = m^2(1-h)$$

$$S_{01} = (1-m)^2 h$$

$$S_{11} = 2mh(1-m) \rightarrow S_{21} = P_2 + Q_3 = \alpha S_{11} - 2\beta S_{21} + \gamma S_{20} - \delta S_{21}$$

$$S_{21} = m^2 h$$

$$\Delta \text{ for example } (m, h) = 2mh(1-m) + m^2 h - 2\beta m^2 h + \gamma m^2(1-h) - \delta m^2 h \\ = 2m(1-m)h - 2\beta m^2 h + \gamma m^2(1-h) - \delta m^2 h \\ = 2mh[\alpha(1-m) - \beta m] + m^2[\gamma(1-h) - \delta h]$$

is clearly satisfied if

$$m = \alpha(1-m) - \beta m$$

$$h = \gamma(1-h) - \delta h$$

& can check others \square

(*)

for r proteins controlling S gates

i proteins $M_1 \dots M_r$ with m_k gates of M_k s.t. $\sum m_k = S$

If $S_{i_1 \dots i_r}$ is density of gates with i_j open M_j gates

and n_j is fraction of open M_j gates then we expect in equilibrium

$$S_{i_1 \dots i_r} = \prod_{j=1}^r \frac{m_j^{i_j}}{C_{i_j}^{i_j}} n_j^{i_j} (1-n_j)^{m_j-i_j}$$

where ${}^m C_i = \frac{m!}{i!(m-i)!}$ is the binomial coefficient.

P.S. 2 answers

4.

H-H model is

$$C_m \dot{V} = I_{app} - I_i$$

$$I_i = g_{Na} m^3 h (V - V_{Na}) + g_K n^4 (V - V_K) + g_L (V - V_L)$$

~~assume~~

$$\begin{aligned} \tau_m &= m_\infty(V) - m \\ \tau_n &= n_\infty(V) - n \\ \tau_h &= h_\infty(V) - h \end{aligned}$$

take V relative to resting potential V_{eq} define $v = V - V_{eq}$, $v_{Na} = V_{Na} - V_{eq}$ etc

$$\text{Scale } v \sim v_{Na}, I_i = s_{Na} v_{Na} g, I_{app} = s_{Na} v_{Na} I^*$$

$$t \sim \tau_n$$

$$\text{then } \varepsilon v = I^* - g, \quad \varepsilon = \frac{C_m}{\tau_n} \frac{s_{Na}}{g_{Na} v_{Na}} = \frac{C_m}{\tau_n s_{Na}}$$

$$g = m^3 h (v-1) + \frac{g_K}{s_{Na}} n^4 (v - \frac{v_K}{v_{Na}}) + \frac{g_L}{s_{Na}} (v - \frac{v_L}{v_{Na}})$$

$$\frac{\tau_m}{\tau_n} \dot{m} = n_\infty - n$$

$$\frac{\tau_m}{\tau_n} \dot{n} = m_\infty - m$$

$$\frac{\tau_h}{\tau_n} \dot{h} = h_\infty - h$$

Now assume $\tau_m \ll \tau_n \Rightarrow m \approx m_\infty(v)$ or just $m(v)$

$$\tau_n = \tau_h, \quad n_\infty + h_\infty = \bar{h}$$

$$\Rightarrow n + h \rightarrow \bar{h}, \quad \text{take } h = \bar{h} - n$$

(2)

then we have

$$i = n_\infty - n$$

$$\epsilon i = I^k - g$$

$$g = \gamma_K (v + v_K^*) n^4 + \gamma_L (v - v_L^*) - (1-v)(\bar{e} - n) m^3(v)$$

$$\gamma_K = \frac{g_K}{g_{Na}} \quad \gamma_L = \frac{g_L}{g_{Na}} \quad v_K^* = -\frac{v_K}{v_{Na}} \quad v_L^* = \frac{v_L}{v_{Na}}$$

$$\gamma_K = \frac{36}{120} \sim 0.3 \quad \gamma_L = \frac{0.3}{120} \sim 0.003$$

$$v_K^* = -\frac{(-12)}{\frac{420}{115}} \sim 0.1 \quad v_L^* = \frac{10.6}{115} \sim 0.1$$

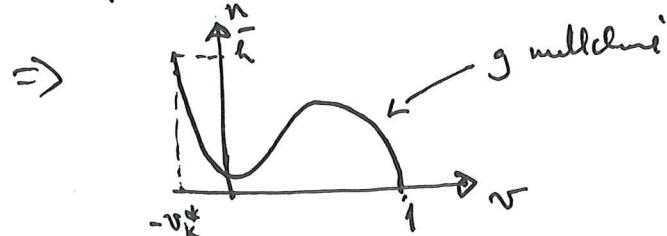
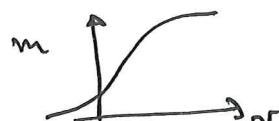
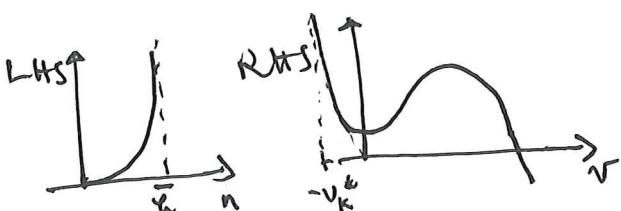
$$\epsilon = \frac{C_m}{T_a g_{Na}} \sim \frac{10^{-6}}{5 \cdot 10^{-3} \cdot 120} \frac{F}{cm^2 s} \frac{am^2}{S} = \frac{1}{600} = 1.6 \times 10^{-3}$$

Form of δ

$\gamma_L \ll 1$, ignore,

$$g \approx 0 \text{ if } \gamma_K (v + v_K^*) n^4 \approx (1-v)(\bar{e} - n) m^3(v)$$

$$\text{or} \quad \frac{\gamma_K n^4}{\bar{e} - n} = \frac{(1-v) m^3(v)}{v + v_K^*}$$



[wouldn't expect this], I didn't do it in lecture]

(3)

[The neglect of γ_L is ok till $n = \bar{n}$. When $n = \bar{n}$,

$$\text{then } \gamma_K(v + v_K^*)^{\bar{n}^4} + \gamma_L(v - v_L^*) = 0$$

$$\Rightarrow (\gamma_K^{\bar{n}^4} + \gamma_L)v = -\gamma_K v_K^* \bar{n}^4 + \gamma_L v_L^*$$

$$\text{if } v = \frac{-\gamma_K v_K^* \bar{n}^4 + \gamma_L v_L^*}{\gamma_K^{\bar{n}^4} + \gamma_L} = -\frac{\left[v_K^* - \frac{\gamma_L v_L^*}{\gamma_K^{\bar{n}^4}} \right]}{1 + \frac{\gamma_L}{\gamma_K^{\bar{n}^4}}} v - v_K^*$$

is dn.

For $n > \bar{n}$ the γ_L term is required,

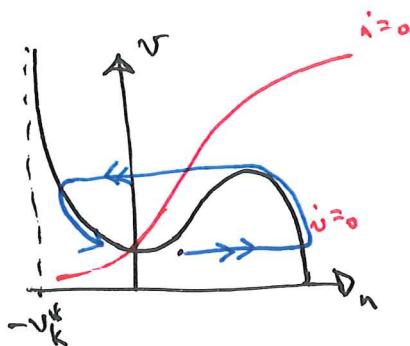
$$\gamma_K(v + v_K^*)^{n^4} + (1-v)(n-\bar{n})m^3(v) \approx \gamma_L(v_L^* - v) > 0$$

In practice m^3 becomes very small ($\sim 10^{-6}$ for $n \approx 0.1$)

$$\text{so in fact } \gamma_K^{n^4} \approx \frac{\gamma_L(v_L^* - v)}{v + v_K^*} \approx \frac{\gamma_L(v_L^* + v_K^*)}{v + v_K^*} \text{ as } v \rightarrow -v_K^*$$

so the asymptote does work]

Thus



If $n(0)$ large enough ($\gg n$ initial clear $v=0$)
fixed point is excitable as shown.

excitable: excursion before returning to rest state.

